# CHEMBIOCHEM

## **Supporting Information**

© Copyright Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, 2008

© Copyright Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, 2008

## CHEMBIOCHEM

### **Supporting Information**

for

Engineering of *Thermus thermophilus* Cytochrome *c*552: Thermally Tolerant Artificial Peroxidase

Hiroshi Nakajima, Yusuke Ichikawa, Yuh Satake, Nobuyuki Takatani, Soumen Kati Manna, Jitumani Rajbongshi, Shyamalava Mazumdar, and Yoshihito Watanabe\*

#### Experimental

**General:** All chemicals were purchased from Nacalai Tesque Co. and Sigma Co. and used without further purification. Deuteride buffers and  $D_2O_2$  solution were prepared according to the methods previously reported.<sup>1</sup>

**Site directed mutagenesis:** Expression and purification of both wild type and variant cytochrome  $c_{552}$  were carried out according to the previous report.<sup>2</sup> The variants were constructed with a QuickChange site-directed mutagenesis kit (Stratagene Co.) by using oligonucleotides 5'-GTACCTCATCCTG<u>GACCTTCTCTACGGCC-3'</u> (for replacement of Val49 with Asp) and 5'-GAAGTACAACGGCGTC<u>GCG</u>TCCTCCTTCG-C-3' (for replacement of Met69 with Ala) and their complementary strands as mutation primers. The underlined sequences correspond to the amino acid residues of mutagenesis.

**Measurements of peroxidase activity:** A reaction mixture (3 mL) containing ferulic acid (200  $\mu$ M) and a protein (0.02  $\mu$ M for HRP, otherwise 0.2  $\mu$ M) in MES-NaOH buffer (20 mM, pH 5.0) was preincubated at each reaction temperature (20 - 80 °C) for 10 min. To initiate reaction, H<sub>2</sub>O<sub>2</sub> (500 mM in H<sub>2</sub>O kept at 25 °C) was added to the solu-

tion at the final concentration of 1 mM. Since ferulic acid provides various oxidation products in the peroxidase reaction, observation of the reaction product is not applicable to the quantitative analysis by UV/Vis spectroscopy.<sup>3</sup> Therefore, the peroxidase activity was monitored with consumption of ferulic acid that shows absorption maximum at 310 nm at pH 5.0 ( $e_{310} = 15.6 \text{ mM}^{-1} \text{ cm}^{-1}$ ).

**Kinetic measurements for association of NH<sub>2</sub>OH:** The reactions of the cytochrome  $c_{552}$  variants (10 µM) with NH<sub>2</sub>OH (500 - 1000 µM) were performed at 25 °C on a Unisoku stopped-flow apparatus RSP-1000 (Unisoku Co.) in 20 mM MES-NaOH buffer (pH 5.0). The kinetic traces at 398 nm that corresponds to the absorption maxima of the variants were used for determining pseudo first-order rates. The association rate constants were given by the slopes of plots of the observed rates versus NH<sub>2</sub>OH concentration.

**Reaction with cumenehydroperoxide:** The reaction mixture containing 10  $\mu$ M cytochrome  $c_{552}$  variants and 300  $\mu$ M cumene hydroperoxide was incubated for 30 min at 25 °C and pH 5.0. Aliquots of the mixture were analyzed by HPLC (Shimadzu Shimpack VP-ODS column). The column was eluted with 50% water / 50% acetonitrile at flow rate of 0.5 mL/min and the effluent was monitored at 210 nm. Assignment of the components was based on the retention time of authentic samples.

**Spectroscopy:** UV/Vis spectra were recorded on MultiSpec 1500 UV/Vis spectrometer equipped with a temperature controller, 110-QS-10 (Shimadzu Co.). Proteins were dissolved into 20 mM MES-NaOH buffer (pH 5.0) at the final concentration of 10 μM. Extinction coefficients of soret bands in the ferric M69A and M69A/V49D variants were determined by pyridine-ferro-hemochrome method to be 161.7 and 146.7 mM<sup>-1</sup> cm<sup>-1</sup>, respectively. Temperature dependent CD spectra were recorded on J-720WN CD spectrometer with a temperature controller (Jasco Co.). A thin optical cell (1 mm path length) containing 200 μL protein solution (10 μM in 20 mM MES-NaOH buffer, pH 5.0) was used for measurements. The temperature was raised lineally from 20 to 95 °C by the ratio of 1 °C/min. All data were corrected at 222 nm that corresponds to a maximum of a negative cotton effect by α-helices.

#### References

- 1 S. Kato, T. Ueno, S. Fukuzumi, Y. Watanabe, *J. Biol. Chem.* **2004**, 279, 52376.
- 2 Y. Ichikawa, H. Nakajima, Y. Watanabe, *ChemBioChem* **2006**, 7, 1582.
- 3 H. Barberousse, O. Roiseux, C. Robert, M Paquot, C. Deroanne, C. Blecker, J. Sci. Food Agric. 2008, 88, 1494.